N-Alkoxycarbonyl Amino Acid *N*-Carboxyanhydrides and *N*,*N*-Dialkoxycarbonyl Amino Acid Fluorides from *N*,*N*-Diprotected Amino Acids†

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Activation of *N*,*N*-bis-Boc or *N*-Boc *N*-Z amino acids with SOCI₂–DMF leads to *N*-protected *N*-carboxy amino acid anhydrides, whereas treatment with cyanuric fluoride at low temperature gives mainly *N*,*N*-bis-protected amino acid fluorides, which are efficient acylating agents.

The renewed interest in the preparation of N-protected amino acid halides¹ and amino acid N-carboxyanhydrides (NCA)² has recently led to the synthesis of interesting new coupling reagents, the stable *N*-alkoxycarbonyl amino acid fluorides (UAAF; Fmoc, Z or Boc derivatives)³ and *N*-alkoxycarbonyl *N*-carboxy amino acid anhydrides (UNCA)⁴, and their use in peptide synthesis has been exemplified.^{3.4} We report here a new route to UNCA **4** and the synthesis of the unknown bis(alkoxycarbonyl) amino acid fluorides (U₂AAF; **3**). Both reagents, which possess easily cleavable *N*-protecting groups, could be useful for acylation of anionic nucleophiles since they are devoid of an exchangeable NH hydrogen.

Depending on the activating agent and experimental conditions, reaction of N, N-bis(alkoxycarbonyl) amino acid 1⁵

[†] *Abbreviations*: Boc: *tert*-butoxycarbonyl; Z = benzyloxycarbonyl; Fmoc = 9-fluorenylmethyloxycarbonyl; Bzl = benzyl; NCA = amino acid *N*-carboxy anhydride; UNCA = urethane-protected amino acid *N*-carboxy anhydride; UAAF = urethane-protected amino acid fluoride; U₂AAF = *N*,*N*-bis(urethane) amino acid fluoride; DMF = dimethylformamide.

Table 1	Table	1
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Stanting	·····			Due de eth		Lit ⁴		
compound	Method ^a	Product	M.p./°C	yield (%)	$[\alpha]_D^c$	M.p./°C	$[\alpha]_{D}^{c}$	
1a	А	3a d	50-52	76	_	_		
1a	В	4a	147-149	72	_	_	_	
1b	А	$\mathbf{3b}^d$	43-45	82	-112.8		_	
					<i>c</i> 0.9, EtOAc			
1b	В	4b	102-104	92	+120.1			
					$c1.8,\mathrm{THF}$			
1c	Α	3c ^d	oil	79	-123.0	<u> </u>		
					c1.2 EtOAc			
1c	В	4c	105-106	75	+138.8	105-106	+127.6	
					c 1.8, THF		c 1.78, THF	
1d	В	4d	101-103	80	+56.9	103-104	+21.6	
					c 1.8, THF		c 1.78, THF	
le	В	4e	117-119	86	+59.7			
					c 1.8, THF			

^{*a*} Method A: (i) **1** + 1 mol equiv. pyridine + 1 mol equiv. cyanuryl fluoride/CH₂Cl₂/-30 °C/90 min; (ii) H₂O; (iii) MgSO₄. Method B: (i) **1** + 1 mol equiv. pyridine + 1 mol equiv. [SOCl₂/DMF]/CH₃CN/20 °C/120 min; (ii) H₂O/EtOAc; (iii) MgSO₄. ^{*b*} Satisfactory elemental analyses were obtained for all compounds. ^{*c*} $[\alpha]_D^{22}$ for compounds **3** and $[\alpha]_D^{25}$ for compounds **4**. ^{*d*} ¹H NMR analysis of crude product shows the presence of small amounts of protected NCA **4**.



gave either the new N, N-bis-protected amino acid fluorides 3, when treated with cyanuric fluoride⁶ (method A), or the urethane protected amino acid N-carboxyanhydride 4, when the SOCl₂/DMF Vilsmeier reagent⁷ (method B) was used instead (Scheme 1 and Table 1). In method A, a trace amount of N-protected NCA 4 is formed beside 3 when working at -30 °C.⁸

N,N-Bis-protected amino acid fluorides **3** are efficient acylating agents. They allowed us to prepare the first examples of fully protected N-acyl derivatives of pyrrole-2-carboxylic acid⁹ (amino acid pyrrolides), namely the N-(N',N'-bis-Bocphenylalanyl)-2-pyrrolecarboxylic acid benzyl ester **5** and the N(N'-Boc N'-Z phenylalanyl)-2-pyrrolecarboxylic acid *tert*butyl ester **6** by treatment with the sodium salts of pyrrole-2carboxylic acid esters. They can also be used for peptide bond formation. For instance Boc₂-Phe-Leu-OBzl^{5b} was obtained from benzyl leucinate in 80% yield. The N-Boc N-carboxy amino acid anhydrides **4** are unreactive towards anions of pyrrole-2-carboxylic acid esters.

N,N-Bis-protected amino acid fluorides **3** are stable compounds. They are easily characterized by ¹⁹F NMR (δ 28–32

relative to $CFCl_3$; coupled to CH_{α}) and by ¹H NMR (CH_{α} coupled to F and at a lower field than the corresponding H_{α} in 1 and 4). They do not spontaneously cyclize to give the corresponding protected amino acid *N*-carboxyanhydrides.

Compounds 4 are probably formed from another activated bis-protected amino acid derivative such as the triazinyl ester 2,¹⁰ the imidoyl ester 2' or the chloride 2''. Participation of a neighbouring *tert*-butyl carbamate is preferred to that of a benzyl carbamate¹¹ probably because of the higher stability of a tertiary carbonium ion: an *N*-Z NCA is obtained from a *N*-Boc *N*-Z amino acid ($1c \rightarrow 4c$; Table 1).

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